1.	(a)	rectally/by suppository, by inhalation, by injection (parenterally), by applying to skin/topically; [2] for three, [1] for two. Award [1 max] if intravenous, subcutaneous and intramuscular are given.	2	
	(b)	an effect produced as well as the one intended/unwanted or undesired effect	; 1	[3]
2.	intra	muscular/into muscles; venous/into veins; utaneous/into fat; (Award [2] for three correct [1] for two or one correct);		
	the d	venous; rug is circulated/transported quickly via the blood stream arious parts of the body); Accept parenteral (other than by mouth).	4	[4]
3.	(a)	hydrochloric acid/HC1;	1	נייו
	(b)	$Mg(OH)_2 + 2HCl \rightarrow MgCl_2 + 2H_2O;$		
		NaHCO ₃ + HCl \rightarrow NaCl + H ₂ O + CO ₂ ; Accept equations with H^+ instead of HCl / OH ⁻ instead of Mg(OH) ₂ and HCO ₃ ⁻ instead of NaHCO ₃ .	2	
	(c)	$Al(OH)_3;$		
		neutralize 0.03 mol of H ⁺ /contains three OH ⁻ ions/OWTTE; Do not award second mark if other than Al(OH) ₃ chosen	2	[5]
4.	(a)	$Mg(OH)_2 + 2HCl \rightarrow MgCl_2 + 2H_2O/Al(OH)_3 + 3HCl \rightarrow AlCl_3 + 3H_2O;$ Award [1] for correct reactants and products and [1] for balancing.	2	[0]
	(b)	Al(OH) ₃ /aluminium hydroxide;	1	
	(c)	corrosive to body/tissue/strong base/alkali;	1	[41
5.	(a)	bacteria;		[4]
		interfere with cell wall formation; prevent formation of cross-links (within wall); size/shape of cell cannot be maintained; water enters the cell/osmosis occurs; cell bursts/disintegrates; Award [1] each for any three of the last five points.	4 max	
	(b)	(overprescription) makes penicillins less effective; they destroy useful bacte allow a resistant poulation to build up/OWTTE;	eria; 3	[7]

6.	(a)	C ₁₆ H ₁₈ O ₄ N ₂ S; Accept elements in any order	1	
	(b)	prevents deactivation by stomach acid/more resistant to stomach acid; prevents deactivation by the enzyme penicillinase (produced by bacteria) /increases resistance/tolerance to penicillinase;	2	
	(c)	penicillin interferes with the cell wall formation; cells can expand/burst/disintegrate/bacteria die;	2	
	(d)	makes penicillin less effective; destroys useful/beneficial bacteria; allows resistant population to build up; Award [1] each for any two.	2	[7]
7.	(a)	bacteria are larger/viruses are smaller; bacteria are cellular/viruses are non-cellular; bacteria have/nucleus/cytoplasm/cell membrane/organelles/ opposite for viruses;		[1]
		bacteria can feed/excrete/respire/grow outside cells/opposite for viruses; Accept "bacteria are living whereas viruses are non-living".		
		viruses insert DNA/RNA into cells/rely on a host cell to reproduce; bacteria multiply by cell division/binary fision/mitosis/meiosis; <i>Award</i> [1] each for any four.	4 max	
	(b)	they alter the host cell's genetic material; they prevent the virus from multiplying; they alter the virus's binding site on the cell wall/they alter the structure of the cell wall to prevent the virus entering; they prevent viruses from leaving the cell; Award [1] each for any two.	2 max	
8.	(a)	bacteria multiply by cell division/binary fission/mitosis; viruses insert DNA/RNA/genetic material into cells; For "bacteria multiply by themselves but viruses require a host cell"/OWTTE award [1].	2	[6]
	(b)	block enzyme activity within host cell/block reverse transcriptase; alter host cell's genetic material; prevent virus from multiplying/replicating; alter virus's binding site on cell wall/prevent virus binding with cell wall; prevent virus from entering/leaving cell; Award [1] each for any two.	2	
	(c)	HIV virus mutates rapidly; HIV metabolism linked to that of host cell/HIV uses host cell; Drugs harm host cell as well as HIV/difficult to target HIV without damagin host cell;	g 2	[6]
9.	(a)	mild analgesics they prevent/interfere with the production of substances/prostaglandins that cause pain; they intercept pain at its source;		[-]
		strong analgesics they bond to receptor sites in the brain; pain signals within brain/spinal cord blocked;	4	

	(b)	(i)	CH ₃ CO/COCH ₃ ; Do not accept ester group/CH ₃ COO.	1	
		(ii)	acetaminophen (paracetamol) amide;		
			hydroxy(l)/phenol/alcohol; ibuprofen carboxylic acid; Do not accept carboxyl.	3	
		(iii)	Ignore any formulas even if wrong. ibuprofen; asymmetric/chiral carbon atom/carbon atom joined to four different groups; Award second mark even if ibuprofen not chosen.	2	[10]
10.	(a)	mild analgesic intercepts pain at the source/OWTTE; by interfering with the production of substances/(enzymes) that cause pain/prostaglandins/OWTTE;			[10]
		binds	g analgesic to pain receptors in the brain; enting the transmission of nerve impulses;	4	
	(b)	(i)	advantage prevents inflammation/thins blood/effective against blood clots/prevents strokes/quick acting/prevents the recurrence of heart attacks/relieves symptoms of arthritis/ rheumatism/reduces fever;		
			disadvantage irritates the stomach lining/produces allergic reactions/ Reye's syndrome/causes stomach bleeding/causes stomach ulcers;	2	
		(ii)	increases the risk of stomach bleeding/haemorrhage/enhances depression of CNS;	1	
		(iii)	may cause kidney/liver damage;	1	
11.	(a)	(i)	antipyretic/reducing fever;	1	
		(ii)	anti-inflammatory/anti-clotting/prevention or treatment of heart attacks/strokes;	1	
	(b)	(i)	ether;		
			alkene/carbon to carbon double bond; (tertiary) amine; Award [1] each for any two.	2	
		(ii)	main effect pain relief;		
			side effect constipation;	2	
12.	(i)	14/14	1.03 (ignore units);	1	
	(ii)		asing amounts needed to produce same effect; asing amounts cause damage/death;	2	[3]

13.	(a)	oxidizing agent/accepts electrons; orange to green;		
	(b)	gas-liquid/chromatography; infra-red spectroscopy;	2	
	(c)	stomach bleeding;	1	[5]
14.	potas			
	orange to green; redox (accept reduction/oxidation);		3	
	1000	(weeep. realieness estimation),	J	[3]
15.	(a)	(tertiary) amine;	1	
	(b)	amide;	1	
	(c)	basic;		
		N atoms can accept H ⁺ ions from water/forms OH ⁻ ions in the solution;	2	
	(d)	anxiety; irritability; sleeplessness;	1	
		increase in urine output;		
		Award [1] for any two. increased blood pressure; reduction in urine output; Award [1] mark for any two.	1	
	(f)	sympathomimetic drug mimics the effect of adrenaline/stimulates the sympathetic nervous system; amphetamine/methamphetamine/speed/ecstasy;	2	101
16.	(a)	both contain six-membered ring; five-membered ring; (tertiary) amine group; N- has methyl group attached; Award [1] each for any two.	2 max	[8]
	(b)	short-term effects increased heart rate/blood pressure/restriction of blood vessels; acts as an anti-diuretic/reduction in urine output;		
		long-term effects increased risk of heart disease/coronary thrombosis; risk of becoming addicted/physically dependent; high cost; (increased risk of) (lung, mouth, throat) cancer; (increased risk of) bronchitis/emphysema; reduction in capacity of blood to carry oxygen; withdrawal symptoms/weight gain (on quitting); Award [1] each for any six, provided at least one short-term effect given.	6 max	

17. (a) amphetamines/stimulants;

> increased heart rate/increased blood pressure/increased breathing rate /dilation of pupils/constriction of arteries/sweating/increased alertness /decreased appetite;

2 1

2

2

2

1

5

2

(b) (i) nicotine;

Accept nicotin.

(ii) increased heart rate:

increased blood pressure;

reduced urine output;

increased concentration/stimulating effect;

Award [1] each for any two.

increased risk of cancer;

increased risk of stroke/(coronary) thrombosis/heart disease;

ulcers;

emphysema/bronchitis/shortage of breath;

coughing/bad breath/yellowing of teeth or fingers;

effect on pregnancy;

Award [1] each for any two.

18. (a) optical;

> chiral/asymmetric carbon atom/carbon joined to 4 different atoms; circle on diagram (around CH joined to N);

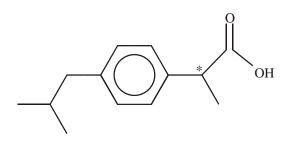
3

alleviates morning sickness; (b)

causes (limb) deformation in fetus;

[5]

19. (a)



(b) a chiral auxiliary is itself an enantiomer;

it is bonded to the reacting molecule to create the stereochemical conditions necessary to follow a certain pathway;

once the desired enantiomer is formed the auxiliary is removed;

different enantiomers may have different biological effects, some of which may be harmful;

genetic defects/deformities/other suitable example;

[6]

20. cause changes in visual and sound perception/hallucinations; (a)

> LSD may cause a permanent effect of "flashbacks"/effects of LSD may be experienced a year or more after the last use of the drug/ psychological dependence;

OR

mescaline may cause nausea/trembling/liver damage/reduce appetite;

(b) both contain amines/aromatic (benzene) ring;

mescaline LSD

primary amine secondary and tertiary amine

ether groups amide
one ring alkene(s)
indole ring
four cyclic rings

heterocyclic 2

Accept any one clear difference from the above list.

[4]

21. (a) (i) cisplatin and geometric/cis-trans isomerism;

Taxol[®] and optical isomerism; 2

Accept other correct examples.

- (ii) atoms/groups arranged differently in space/OWTTE; chiral/asymmetric carbon atom/carbon joined to 4 different atoms/groups;
- (b) chiral auxiliary attaches to starting molecule; chosen reagents convert starting molecule into only one enantiomer; chiral auxiliary removed to leave desired enantiomer; chiral auxiliary is itself optically active/possess a chiral atom; 2

Award [1] each for any two.

[6]

2

22. one enantioner has beneficial/desired effect;

the other enantiomer no effect/harmful effect/waste of material/more clinical trials necessary; thalidomide:

one thalidomide enantiomer relieves symptoms of morning sickness while the other isomer can cause birth defects;

Accept alternatives, e.g.

ibuprofen;

one enantiomer much more effective;

taxol:

one enantiomer much more effective;

[4]

- 23. (a) asymmetric/chiral carbon atom/4 different groups around carbon atom/
 - $\begin{array}{c} \text{HO} & \overset{\star}{ } \text{CH} \text{CH}_2 \text{NH} \text{CH}_3 \\ \text{OH} & \end{array};$
 - (b) one correct 3-D structure;

second structure clearly shown as correct isomer;

2

1

e.g.
$$\begin{array}{c|c} CH_2NHCH_3 & CH_2NHCH_3 \\ & & & \\ \hline \\ C''''''' & H & HO & X \\ \end{array}$$

Accept diagrams that make it clear they are mirror images even though not perfect 3-D representations but must have the chiral carbon at the centre of the molecule.

(c) (i) covalent and coordinate/dative (covalent); square planar; 90°;

3

(ii) 1

[7]